

# CHM 457, CHM1300: Introduction to Polymer Chemistry Fall 2021 Course Syllabus

## I CONTACTS

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### INSTRUCTOR

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Office hours: TBA



### INSTRUCTOR

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### TEACHING ASSISTANT

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Online student hours: by email appointment

## II COURSE OVERVIEW

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### COURSE DESCRIPTION:

This course is an introduction to polymer chemistry. The first part of the course will describe the shapes and sizes of polymer molecules, molecular weight determination, polymer properties in solution and in the solid state. This part will also include phase separation and self-assembly. The second part of the course will focus on polymer synthesis with an emphasis on polymerization mechanisms and on controlling end groups. Examples will illustrate selected applications.

### STUDENT LEARNING OUTCOMES:

By the end of this course, students will be able to:

- identify and describe different polymer structures and understand how the chemical structure affects the properties of a polymer in solution and in the solid state.
- understand different techniques used to characterize polymer materials in the solid state and in solution.
- understand different techniques to determine molecular weights and molecular weight distribution of polymers
- use Excel or other spreadsheet programs to fit osmotic pressure data or polymer viscosity data to determine molecular weights.
- understand various polymerization mechanisms and how they can be used to construct polymers of different compositions.

**PREREQUISITE COURSE(S) for CHM457:** CHM220H1/CHM222H1, CHM247H1/CHM249H1. CHM325H is recommended.

Both courses assume that you have a fundamental understanding of basic organic chemistry and organic reaction mechanisms; and basic physical chemistry (including thermodynamics and reaction kinetics).

#### READINGS:

Required: Painter P.C.; Coleman, M.M. *Essentials of Polymer Science and Engineering* (ISBN 978-1-932078-75-6).

Supplemental: Young, R.L.; Lovell, P.A. *Introduction to Polymers*, 2<sup>nd</sup> ed.

Sperling, L.H., *Physical Polymer Science*, 2<sup>nd</sup> or 3<sup>rd</sup> ed.

### III HOW THE COURSE IS ORGANIZED

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**The first 2 lectures will be delivered online. All other lectures will be delivered in class (lecture room will be announced soon). Tutorials will be delivered online (until further announcement).**

Classes: **Tuesdays 5 - 7 pm EST. The first 2 lectures will be posted on MyMedia**

Tutorials: **Fridays 2 - 3 pm EST, online**

A list of course topics is presented at the end of this syllabus.

#### TUTORIALS

Early tutorials will demonstrate how to use online programs to draw chemical structures, as well as how to fit data using Excel and how to transfer plots from Excel to PowerPoint and to Word. Later tutorials will be devoted to answering questions from the students about material presented in class.

### IV EVALUATION/GRADING SCHEME

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#### RELEVANT SESSIONAL DATES AND MARKING SCHEME:

| Dates                           | Assignments      | Weight |
|---------------------------------|------------------|--------|
| Oct 25                          | Term test 1*     | 15/25% |
| Dec 07                          | Term test 2*     | 15/25% |
| Sept. 28, Oct 13, 27, Nov 9, 23 | Problem Sets (5) | 20%    |
| Due Friday Nov 30, by 5 pm      | Term paper       | 40%    |

The higher term test mark 25%, the lower term test mark 15%

Problem sets are to be submitted on Quercus by 10 am (EDT/EST) on the

**Monday following the assignment.** (See separate file with upload instructions.)

\*One -hour term tests will be held in class. Calculators will be allowed.

**Note:** if an unexpected technical issue occurs with a university system (e.g., Quercus services, network outage) that affects availability or functionality, it may be necessary to revise the timing or weighting of the assessments.

## V COURSE POLICIES

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- Communication with the instructor: students should email from their U of T email account. Please allow 24 hours for a response on **weekdays**; generally, I will not respond on weekends.
- All online interactions are expected to conform with the University statement regarding a positive learning environment: *“The University of Toronto is committed to equity, human rights and respect for diversity. All members of the learning environment in this course should strive to create an atmosphere of mutual respect where all members of our community can express themselves, engage with each other, and respect one another’s differences. U of T does not condone discrimination or harassment against any persons or communities.”*
- Privacy and appropriate use of course materials: Classes and tutorials will be recorded, and the recordings made available for a limited time through Quercus. These recordings should be considered private and for the use of students registered in the course **only**. They may not be shared or reposted in any way. Students may not make their own recordings, either for personal use or distribution. Students with accessibility requirements should contact the course instructor to make appropriate arrangements.
- Late policy for all term work submissions: a penalty of 5% of the maximum mark per day (**10% for problem sets**) will be deducted daily for term work submitted past the deadline unless:
  - An extension had been agreed to by the instructor prior to the deadline
  - An emergency arose and any notification/documentation was completed in accordance with the Rules and Regulations as posted in the Faculty of Arts & Science Calendar
- Submission methods and deadlines: students are responsible for checking the time, date, and submission process for each item of term work. Generally, submissions will be made through Quercus.

Students are responsible for notifying the course instructor of any absences; accommodation for missed term work will be considered on a case-by-case basis.

## VI TECHNOLOGY REQUIREMENTS

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Specific guidance from the U of T Vice-Provost, Students regarding student technology requirements is available here:

<https://www.viceprovoststudents.utoronto.ca/covid-19/tech-requirements-online-learning/>

Advice for students more broadly regarding online learning is available here:

<https://onlinelearning.utoronto.ca/getting-ready-for-online/>

This course requires the use of computers, and of course, sometimes things can go wrong when using them. You are responsible for ensuring that you maintain regular backup copies of your files, use antivirus software (if using your own computer), and schedule enough time when completing an assignment to allow for delays due to technical difficulties. Computer viruses, crashed hard drives, broken printers, lost or corrupted files, incompatible file formats, and similar mishaps are common issues when using technology, and are not acceptable grounds for a deadline extension.

## VII INSTITUTIONAL POLICIES AND SUPPORT

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### ACADEMIC INTEGRITY

#### **On Academic Integrity:**

Academic integrity is essential to the pursuit of learning and scholarship in a university, and to ensuring that a degree from the University of Toronto is a strong signal of each student's individual academic achievement. As a result, the University treats cases of cheating and plagiarism very seriously. The University of Toronto's Code of Behaviour on Academic Matters

(<https://governingcouncil.utoronto.ca/secretariat/policies/code-behaviour-academic-matters-july-1-2019>) outlines the behaviours that constitute academic dishonesty and the processes for addressing academic offences. Potential offences include, but are not limited to:

In papers and assignments:

1. Using someone else's ideas or words without appropriate acknowledgement.
2. Submitting your own work in more than one course without the permission of the instructor.
3. Making up sources or facts.
4. Obtaining or providing unauthorized assistance on any assignment.

On tests and exams:

1. Using or possessing unauthorized aids.
2. Looking at someone else's answers during an exam or test.

3. Misrepresenting your identity.

In academic work:

1. Falsifying institutional documents or grades.
2. Falsifying or altering any documentation required by the University.

All suspected cases of academic dishonesty will be investigated following procedures outlined in the Code of Behaviour on Academic Matters. If you have questions or concerns about what constitutes appropriate academic behaviour or appropriate research and citation methods, you are expected to seek out additional information on academic integrity from your instructor or from other institutional resources (see <https://www.academicintegrity.utoronto.ca/>).

### **Use of Turnitin**

Normally, students will be required to submit their course essays to Turnitin.com for a review of textual similarity and detection of possible plagiarism. In doing so, students will allow their essays to be included as source documents in the Turnitin.com reference database, where they will be used solely for the purpose of detecting plagiarism. The terms that apply to the University's use of the Turnitin.com service are described on the Turnitin.com web site.

### **ACCESSIBILITY NEEDS**

Students with diverse learning styles and needs are welcome in this course. The University of Toronto is committed to accessibility: if you require accommodations for a disability, or have any other accessibility concerns about the course, please contact [Accessibility Services](#) as soon as possible.

### **ADDITIONAL SERVICES and SUPPORT**

The following are some important links to help you with academic and/or technical service and support

- General student services and resources at [Student Life](#)
- Full library service through [University of Toronto Libraries](#)
- Resources on conducting online research through [University Libraries Research](#)
- Resources on academic support from the [Academic Success Centre](#)
- Learner support at the [Writing Centre](#)
- Information for [Technical Support/Quercus Support](#)

## VIII LIST OF TOPICS TO BE COVERED

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### **1. Introduction**

Classification of polymers  
Overview of polymer structure, composition, architecture  
General concepts of molecular weight distribution

### **2. Experimental determination of the shapes and sizes (molecular weights) of macromolecules**

Conformation of macromolecules  
End-group analysis (especially NMR)  
Osmometry  
Scattering techniques  
Viscometric techniques

### **3. Polymer solutions**

Polymer interactions in solutions; concentration regimes, solvent quality  
Solution and melt thermodynamics (Flory-Huggins theory)  
Solubility parameter approach  
Association in polymer solution and melts  
Phase separation

### **4. Polymers in the bulk state**

Melts and amorphous states  
Glass transition  
Crystalline polymers  
Morphology  
Macroconformation, packing, chain folding  
Kinetics of crystallization  
Mechanical properties of polymers.

### **5. Polymer synthesis: anionic, cationic, and ring opening living polymerization**

Polymerization of styrene, methacrylates, tetrahydrofuran, ethylene oxide.  
Block copolymer synthesis

### **6. Polymer synthesis: Free radical polymerization**

Kinetics and mechanism  
Controlling  $M_n$ , chain transfer  
Copolymerization and reactivity ratios

### **7. "Living/controlled" Radical polymerization**

Nitroxide mediated polymerization (NMP, also called SFRP)

Reversible addition-fragmentation transfer (RAFT) polymerization  
Atom transfer radical polymerization (ATRP)  
Comparison of “living/controlled” radical polymerization and free radical polymerization

**8. Step-growth polymerization**

Mechanism and statistics of step-growth polymerization  
Gel formation

**9. Polymer synthesis: other topics**

Ring-opening metathesis polymerization